



# Title: Biosynthesis of Nano Metallic Particles

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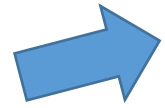
# Introduction

In recent years nanomaterials have been of great interest to the scientific-technological community for presenting physical, chemical, electronic and magnetic properties different from their macroscopic structures; their dimensions are in a range of the nanoscale (less than 100 nanometers), have a larger surface area, making these materials more active, their quantum effects can produce significant changes in their properties compared to their volume analogues. However, the indiscriminate use of batteries and batteries has caused a great problem at the environmental and health level due to its high level of pollution in soils, water and air, which causes irreversible damage to the ecosystem, as in human health since some components of the batteries can cause damage to the respiratory system, kidneys, cancer of all kinds, etc. Therefore, this research used the biomass of the aquatic lily (*Eichhornia Crassipes*) as abioresponsive agent, in order to obtain an ecological and economic method in obtaining nanoparticles of ZnO and MnO using discarded batteries of Zinc-carbon type (AA) since there are few methods of recovery of battery waste that are ecological and that provide an important step towards the synthesis of these.

# Methodology

## SAMPLE EXTRACTION

**BATTERY  
COLLECTION**



**Stacks were  
opened manually**



**Component  
separation**



**Plastic**

**Paper**

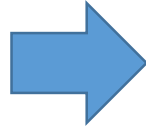
**casing**

**Graphite rod**

**Electrolyte**

# SAMPLE PREPARATION

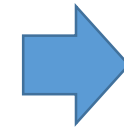
The electrolyte samples  
(black paste) were allowed  
to dry at room temperature



1st reading  
on Rx

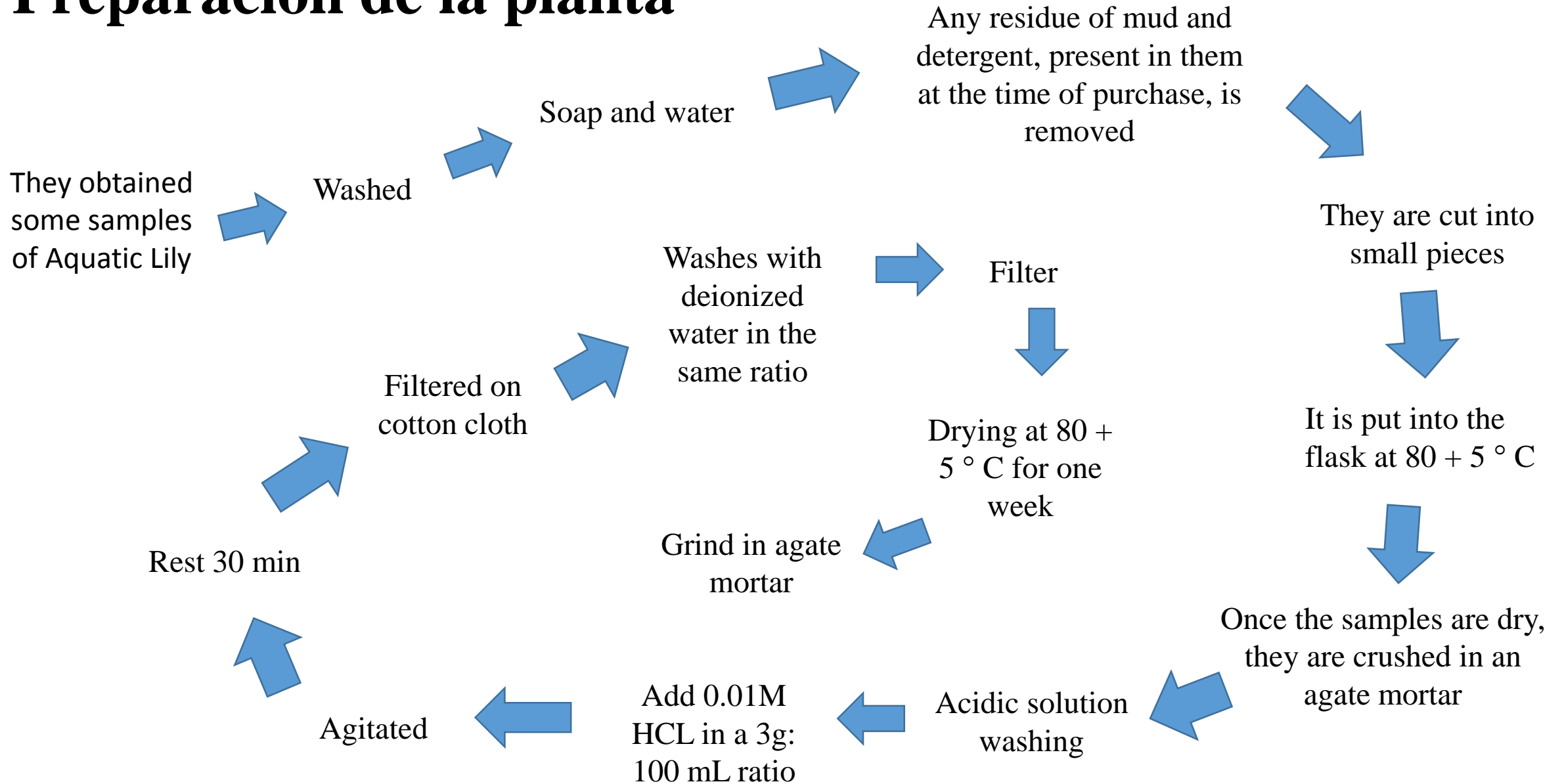


36 hrs in the muffle  
at 800 ° C

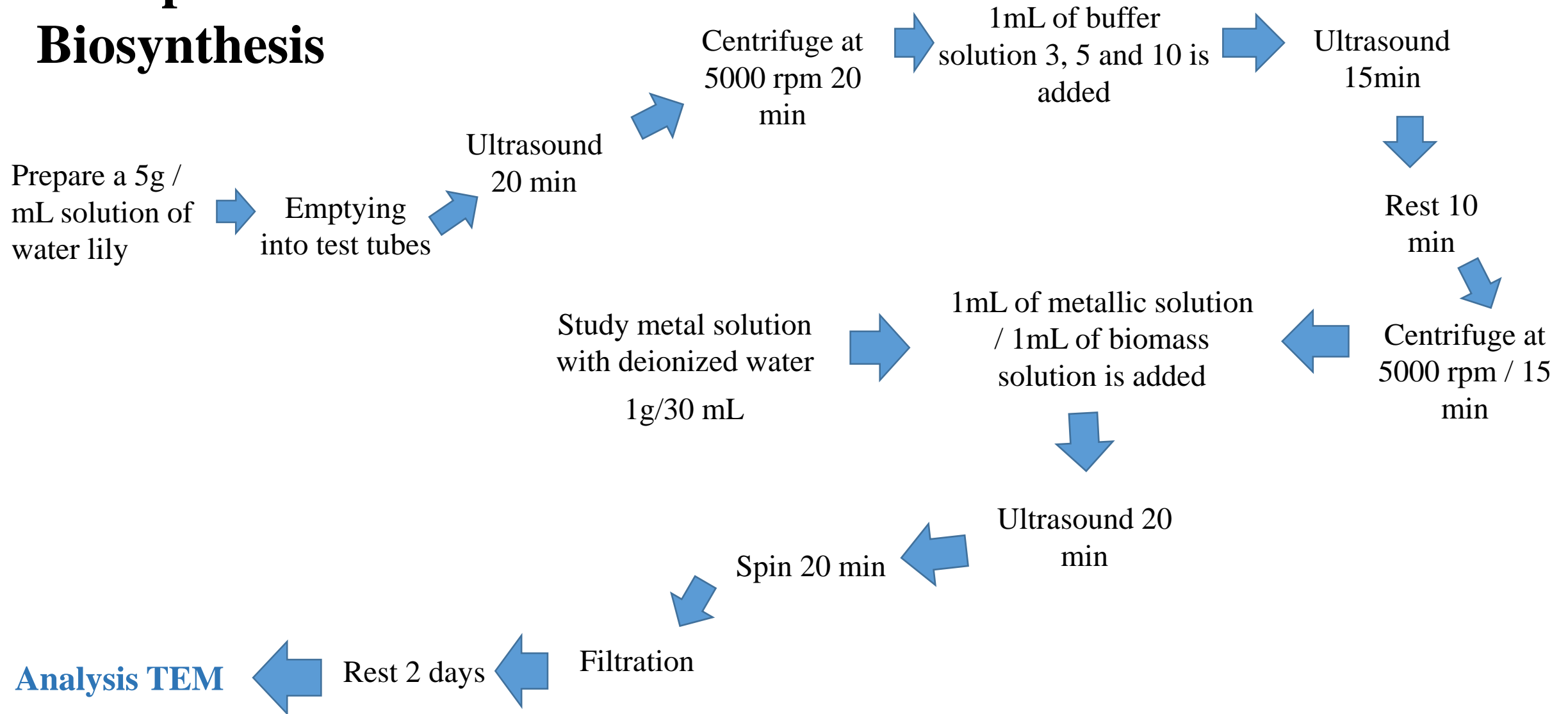


More defined phases  
of the components  
are obtained

# Preparación de la planta

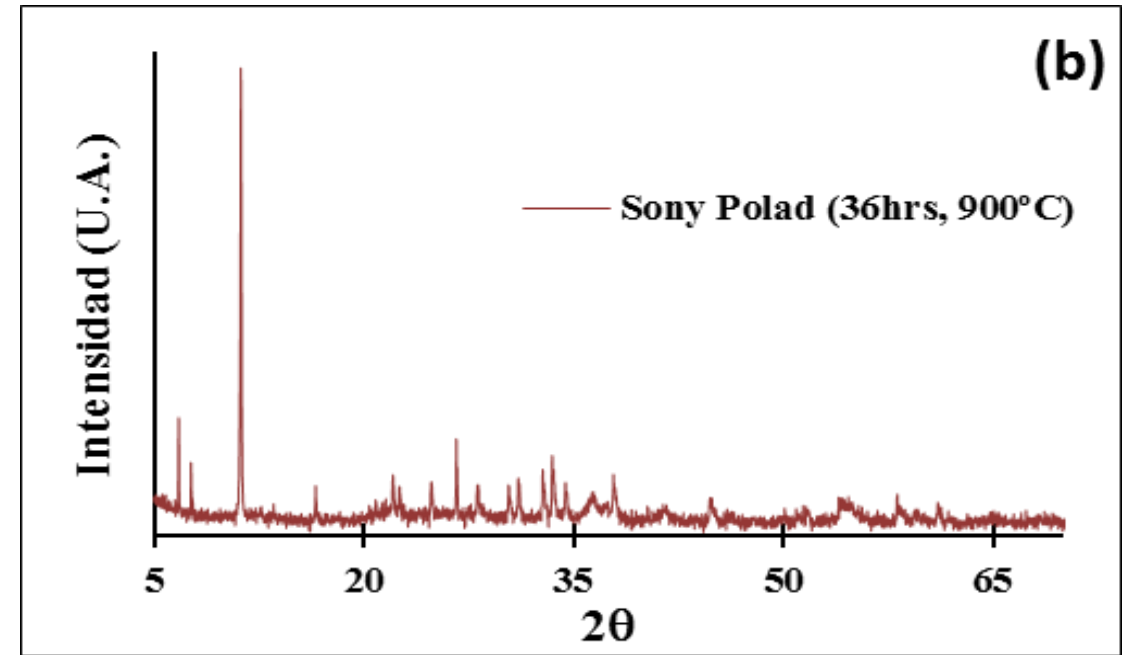
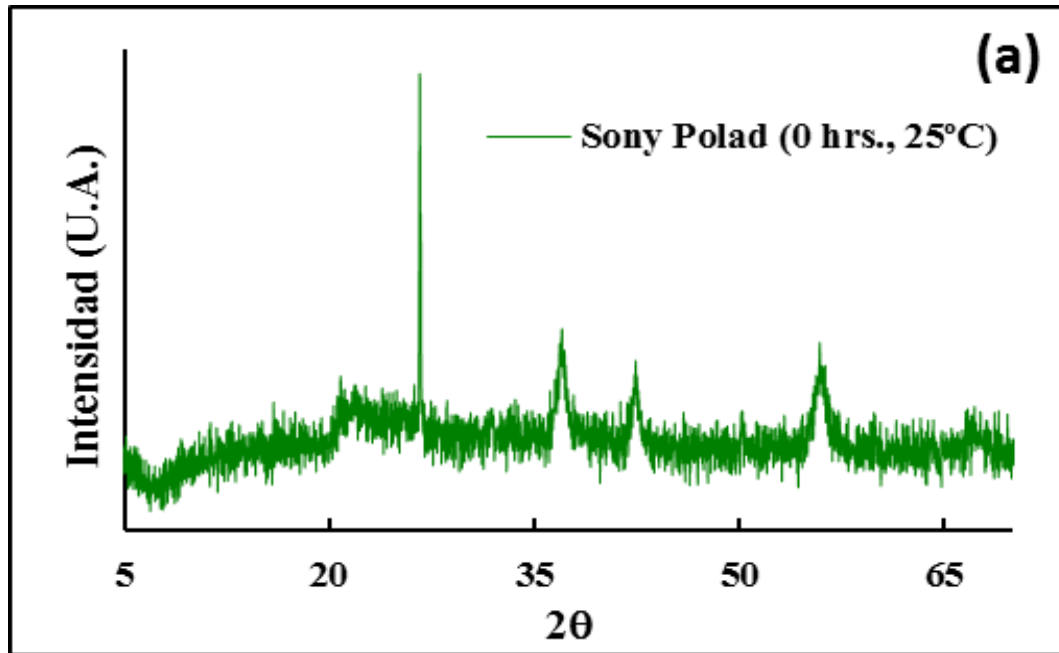


# Nanoparticle Biosynthesis

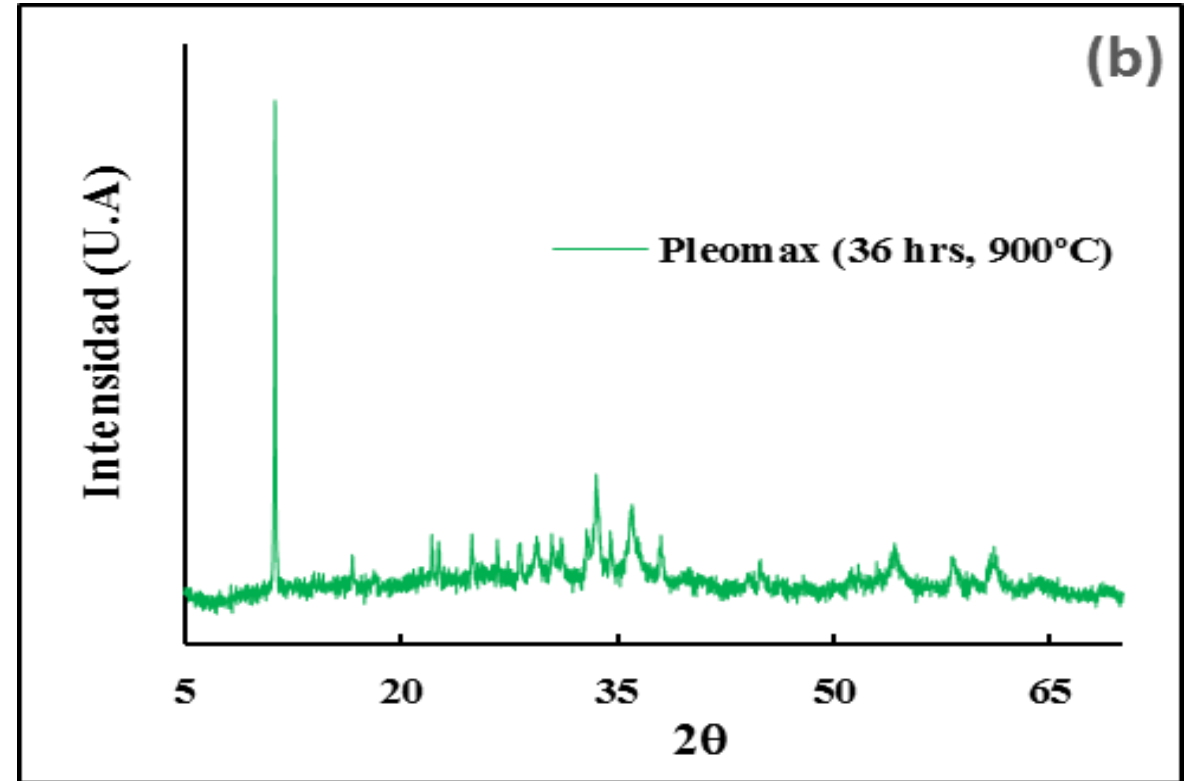
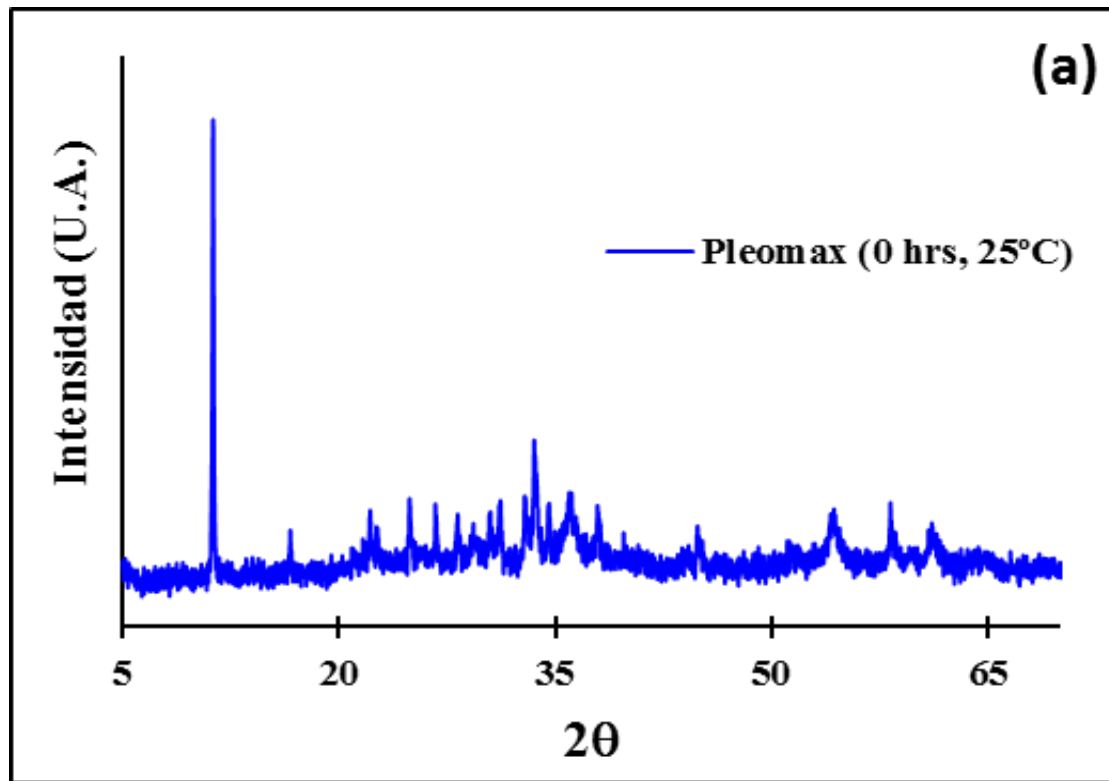


# Results

## XRD Diffraction Study



**Figure 1.1 X-ray diffraction spectra, obtained in the Sony Polad stack (SP1) at different: a)  $T = 25^{\circ}\text{C}$  and  $t = 10$  hrs; b)  $T = 900^{\circ}\text{C}$  and  $t = 36$  hrs**



**Figure 1.2 X-ray diffraction spectra, obtained in the Pleomax stack (PLx) at different: a)  $T = 25^{\circ}\text{C}$  and  $t = 0$  hrs; b)  $T = 900^{\circ}\text{C}$  and  $t = 36$  hrs**



# Transmission Electron Microscopy Study with TEM- microanalysis

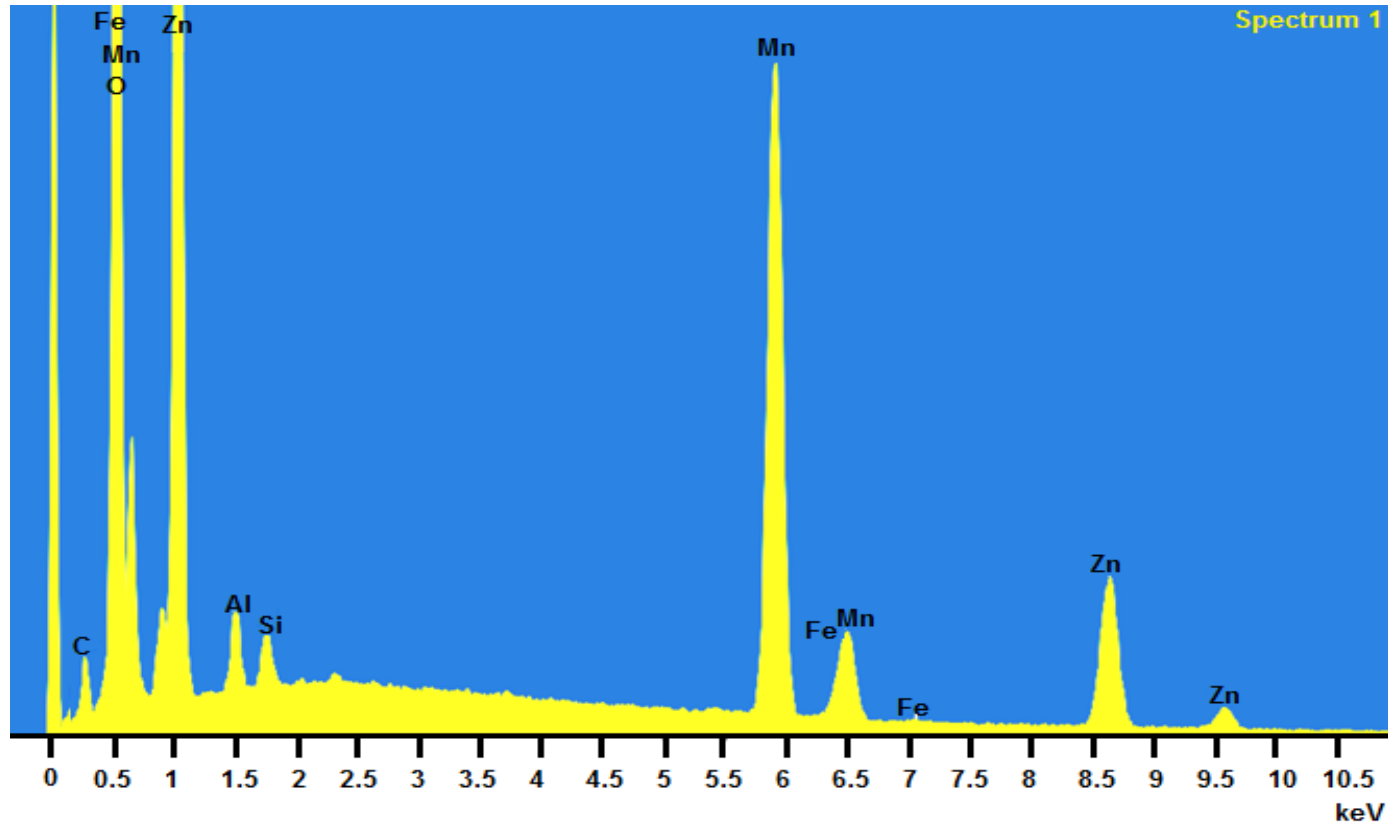


Figure 3. SEM-EDS of the sample (SP1)

Table 1.1 Chemical composition of the sample paste (SP1)

Element	Weight%	Weight%	Atomic%
		Sigma	
C	3.16	0.15	8.22
O	28.16	0.14	54.95
Al	0.92	0.03	1.06
Si	0.58	0.03	0.64
Mn	32.48	0.15	18.46
Fe	1.19	0.10	0.66
Zn	33.52	0.16	16.01
<b>Totales</b>	<b>100.0</b>		

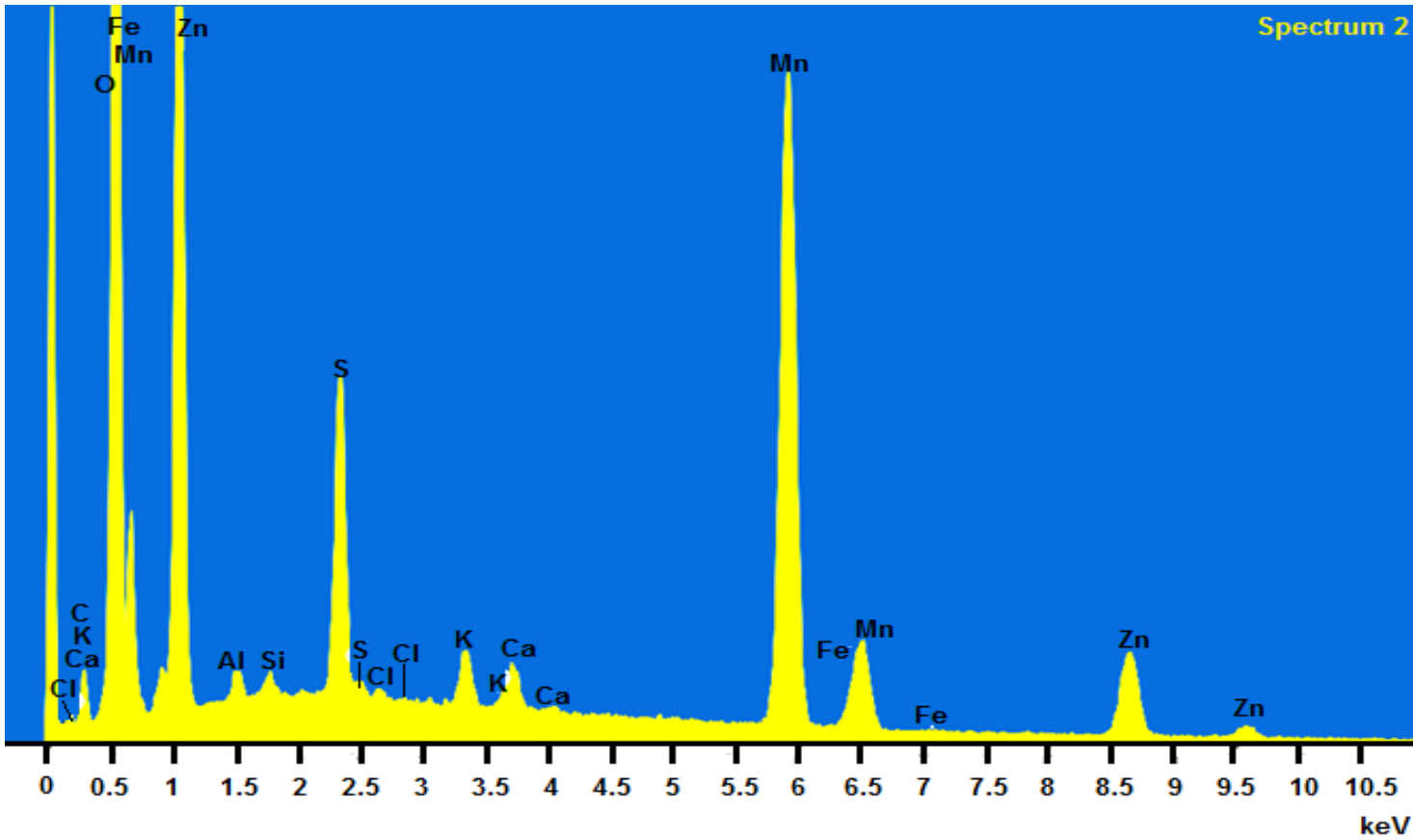


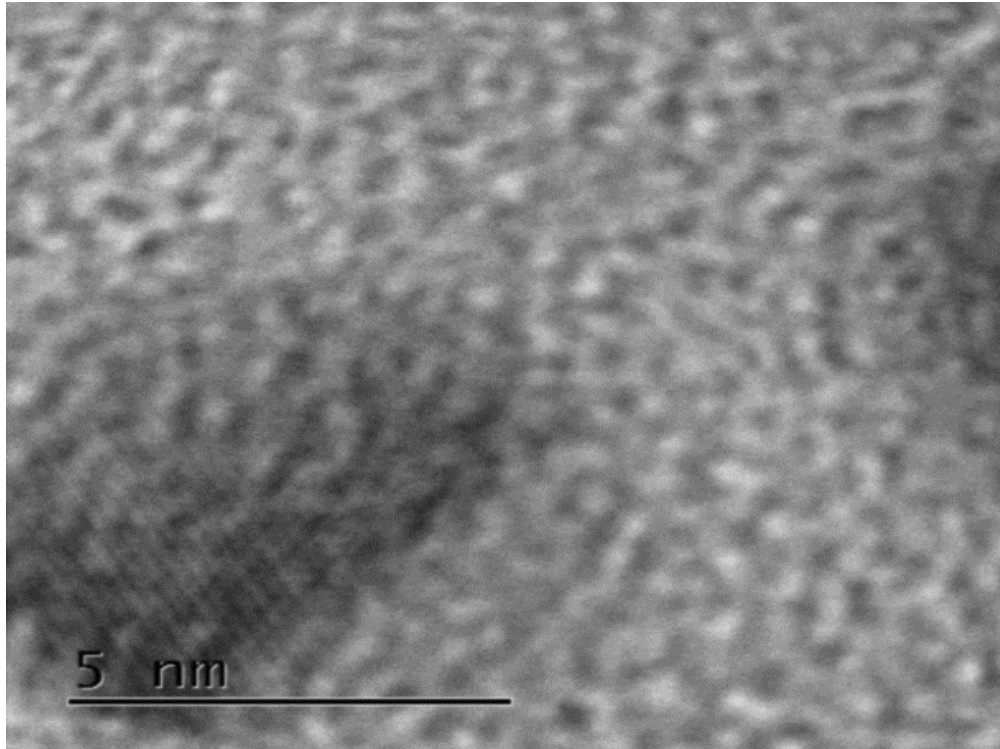
Figure 4. TEM of the sample (PLx)

Table 1.2 Chemical composition of the sample paste (PLx)

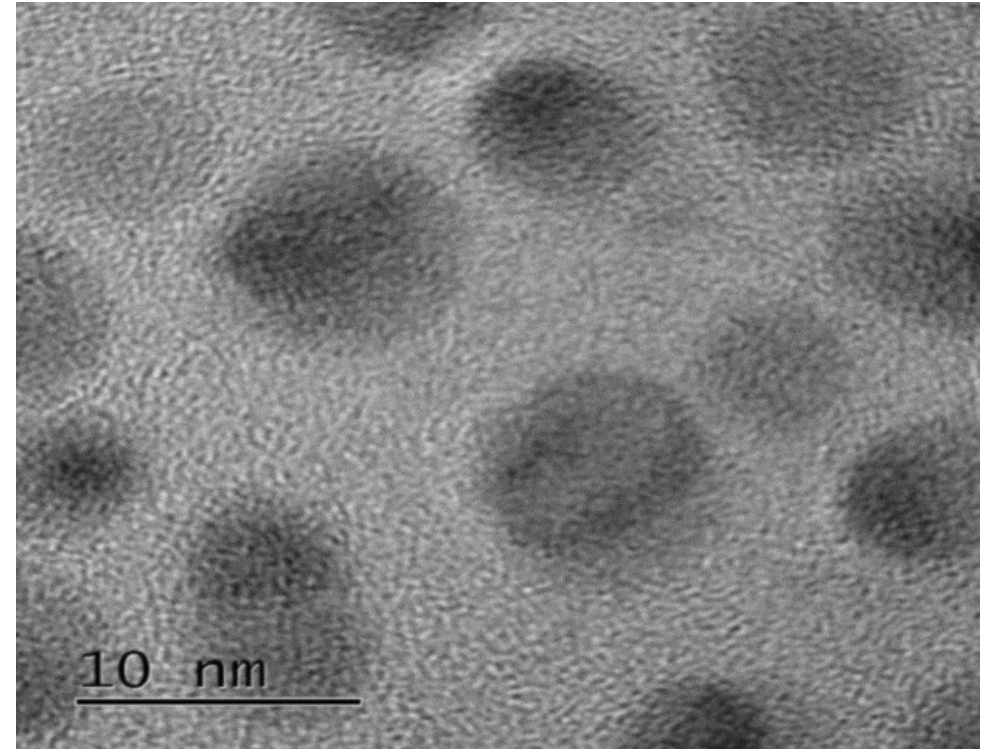
Element	Weight%	Weight%	Atomic %
		<b>Sigma</b>	
C	3.11	0.24	7.54
O	31.55	0.17	57.50
Al	0.36	0.03	0.39
Si	0.21	0.03	0.22
S	3.93	0.05	3.58
Cl	0.16	0.03	0.13
K	0.92	0.03	0.68
Ca	0.77	0.03	0.56
Mn	35.41	0.18	18.79
Fe	1.07	0.10	0.56
Zn	22.51	0.17	10.04
<b>Total</b>	<b>100.00</b>		

# Results of the Biosynthesis of nanometals, using the Transmission Electron Microscopy (TEM) technique

Image 1.1 shows the mapping obtained by TEM of the SP1 sample at a pH value = 3 established for bioreduction, on a scale of: a) 5nm and b) 10 nm.

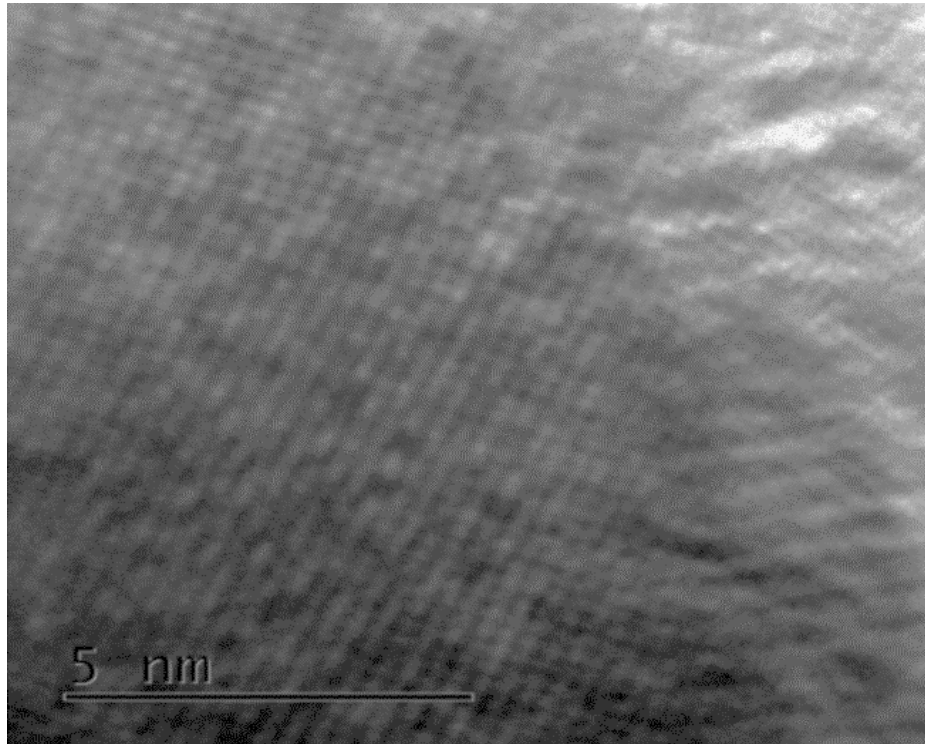


**Imagen 1.1 a) pH=3, Escala: 5 nm**

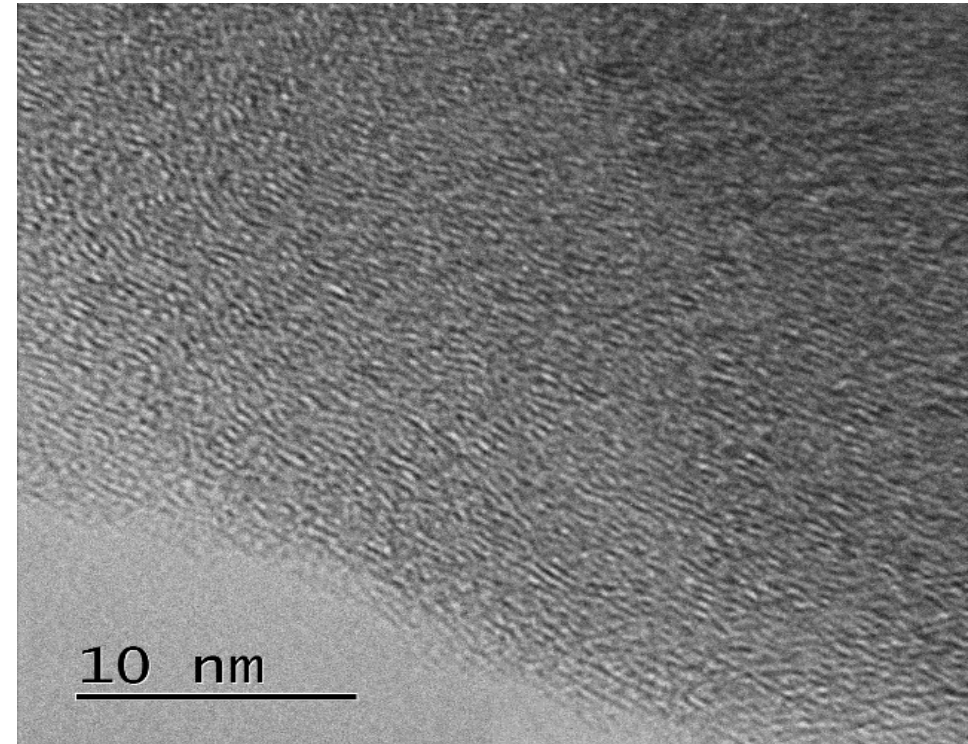


**Imagen 1.1. b) pH=3, Escala: 10 nm**

Image 1.2 shows the image obtained from the mapping carried out by TEM of the PLx sample at a pH value = 3 established for bioreduction, on a scale of: a) 5nm and b) 10 nm.



**Imagen 1.2a) pH=3, Escala: 5nm**



**Imagen 1.2b) pH=3, Escala: 10nm**

# Conclusions

The metallic nanoparticles of ZnO and MnO were obtained in a satisfactory reproducible manner from the discarded batteries of the Sony Polad and Pleomax brands, using the water lily as a bioreducer.

The X-ray diffraction studies allowed to verify the presence of the expected main phases, MgO and ZnO and that there is a better crystallization of the components of the batteries used when they were subjected to a temperature of 900 ° C in a time of 36 hours since they show a better definition in the signals and noise reduction

By means of TEM it was possible to identify the composition of the batteries used and it was observed that they show elements that are not yet reported in the literature, which is clear that some manufacturers use catalysts in order to have a better performance, without considering that at Once discarded, the batteries end up in open-air garbage dumps and the components react, contaminating the soil, water, animals, etc. in an excessive and irreversible way

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