



**Title: Epazote solar drying under different conditions: kinetics, modeling, and colorimetry**

**Authors: CASTILLO-TÉLLEZ, Beatriz, CASTILLO-TÉLLEZ, Margarita, MEJÍA-PÉREZ, Gerardo Alberto and VEGA-GÓMEZ, Carlos Jesahel**

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 143 – 50 Itzopan Street  
 La Florida, Ecatepec Municipality  
 Mexico State, 55120 Zipcode  
 Phone: +52 1 55 6159 2296  
 Skype: ecorfan-mexico.s.c.  
 E-mail: contacto@ecorfan.org  
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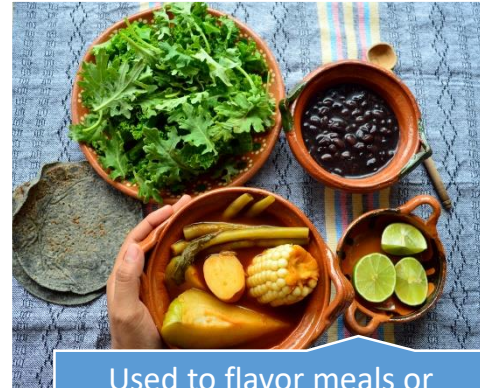
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# Introduction



Epazote is a widely used plant in Mexico



Used to flavor meals or infusions in traditional medicine



It is a plant rich in flavonoids and natural antioxidants



It has very high moisture content



Perishable



The drying of this plant is proposed



# Introduction



Industrial drying requires a high energy consumption, using gas or electricity, promoting climate change and high costs



Open solar drying is a very old technology, is economic and simple method



Presents problems such as pollution, exposure to weather conditions, long drying times





# Methodology

It is proposed to use a solar dryer and test two different types of materials to compare their efficiencies

The materials used are

Titanium oxide

Polyethylene





## Methodology

- Drying kinetics
- Inside and ambient temperatures
- Moisture content and
- Colorimetric study were monitored



# Methodology

A comparison of drying kinetics with existing mathematical models in literature was made

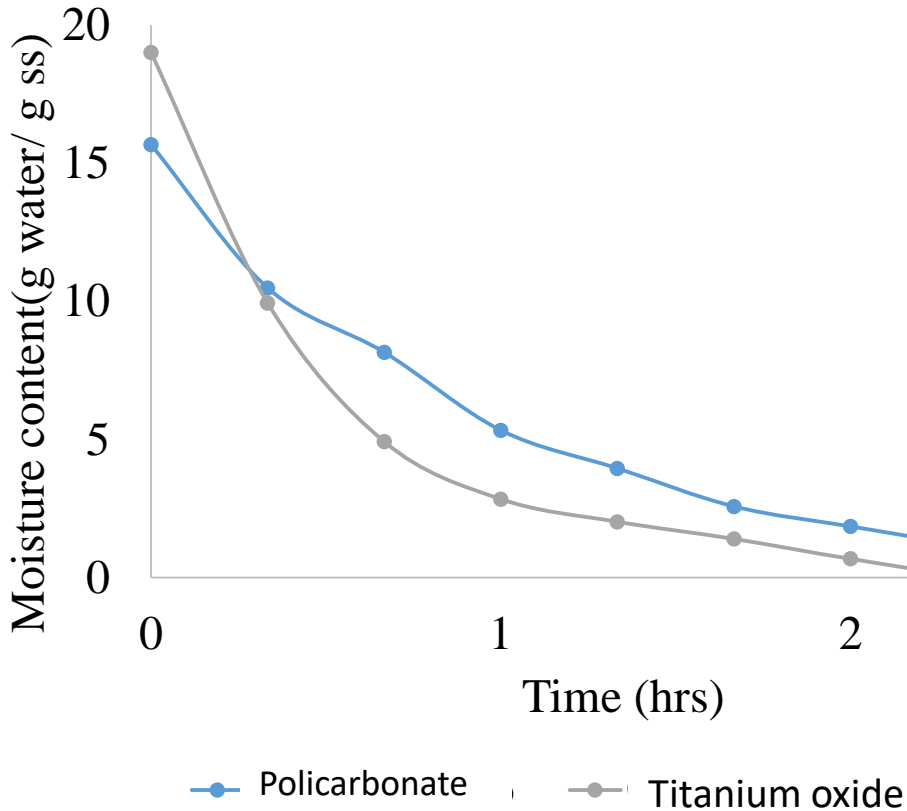
Model	Equation	Reference
Newton	$MR = \exp(-kt)$	(Tunde-Akintunde, 2011)
Page	$MR = \exp(-kt^n)$	(Page, 1949)
Modified page	$MR = \exp((-kt)^n)$	(Diamante & Munro, 1993)
Henderson and Pabis	$MR = a \exp(-kt)$	(Henderson & Pabis, 1961)
Logarithmic	$MR = a \exp(-kt) + c$	(Togrul & Pehlivan, 2002)
Two-term	$MR = a \exp(-kt) + b \exp(-k_0t)$	(Koua et al., 2009)
Two-term exponential	$MR = a \exp(-kt) + (1 - a)$	(Y. I. Sharaf-Eldeen et al., 1980)
Wang and Singh	$MR = 1 - at - bt^2$	(Wang & Singh, 1978)
Weibull	$MR = \exp(-(t/b)^\alpha)$	(Midilli et al., 2002)



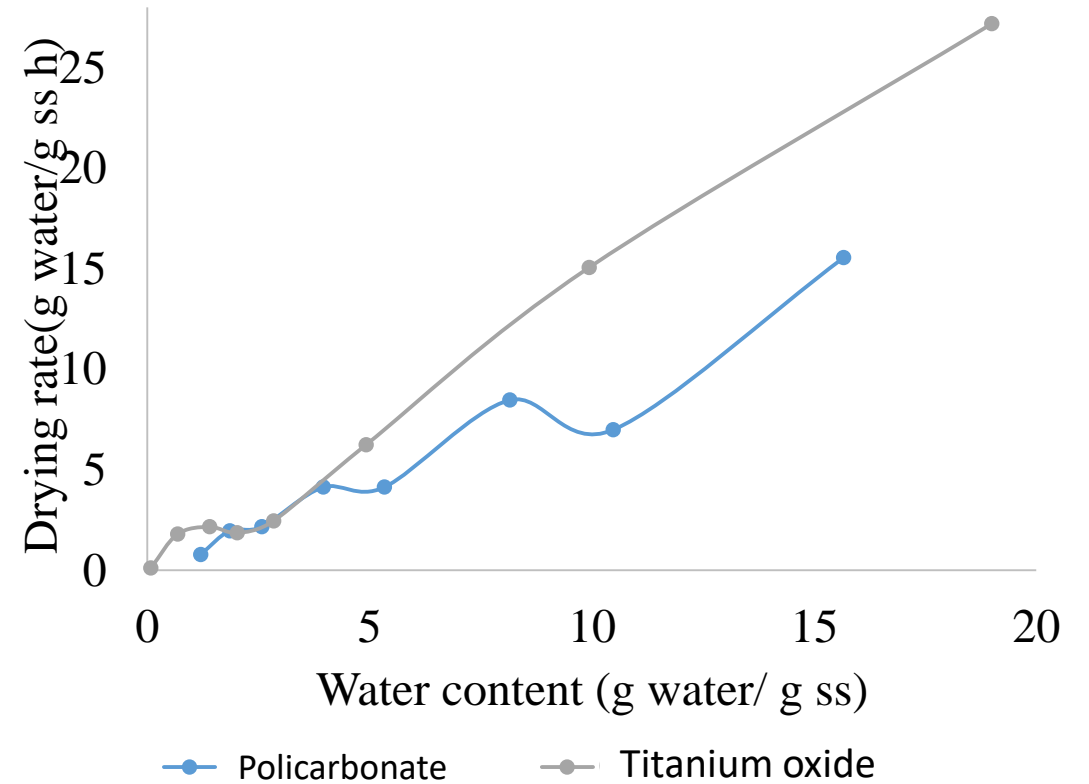


# Results

## Drying kinetics



## Drying rate



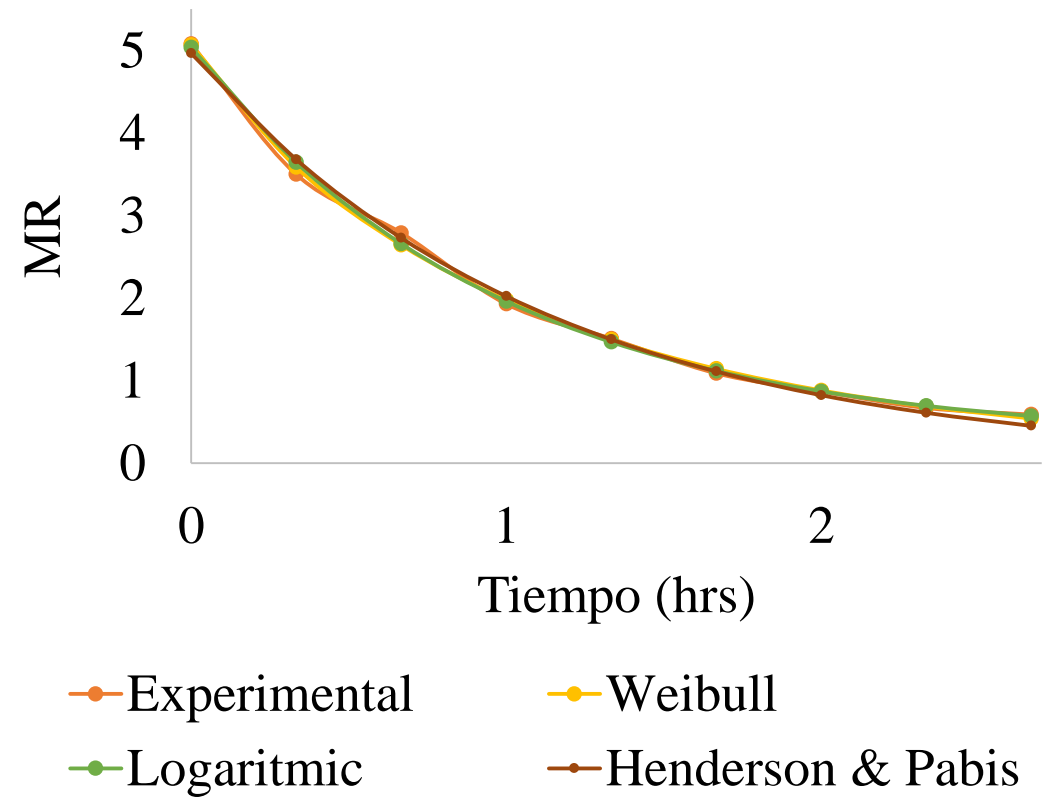
The final values of  $a_w$  for the samples were 0.597 for the titanium oxide cover and 0.533 for the polycarbonate dryer



# Results

## Adjustment parameters with the polycarbonate dryer

Weibull	R <sup>2</sup>	0.9979
	a	8.19E-02
	b	-4.9855
	k	0.9667
	n	0.91974
Logarithmic	R <sup>2</sup>	0.9975
	a	4.7644
	c	0.2704
	k	1.0350
	Henderson and Pabis	R <sup>2</sup>
a		4.9657
k		0.8976

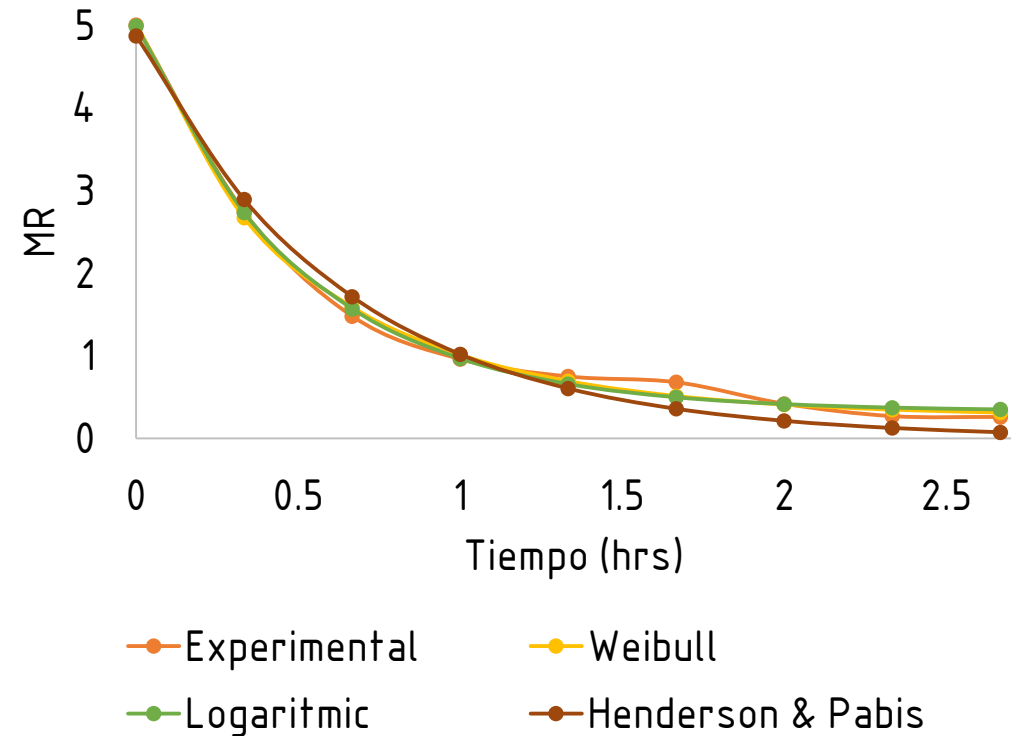




# Results

## Adjustment parameters with the titanium oxide dryer

<b>Weibull</b>	<b>R<sup>2</sup></b>	<b>0.9969</b>
	a	2.60E-01
	b	-4.7484
	k	1.8404
	n	0.9097
<b>Wang and Sing</b>	<b>R<sup>2</sup></b>	<b>0.9964</b>
	a	4.6619
	c	0.3274
		1.98E+0
	k	0
<b>Henderson and Pabis</b>	<b>R<sup>2</sup></b>	<b>0.9835</b>
	a	4.8716
	k	1.5647



# Results



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Polycarbonate dryer

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$\Delta L = -3.89$ ,  $\Delta a = 17.50$ ,  $\Delta b = -9.10$ ,  
 $\Delta C = -10.49$ ,  $\Delta H = -16.70$ ,  $\Delta E = 20.11$

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Titanium oxide

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$\Delta L = -2.78$ ,  $\Delta a = 5.90$ ,  $\Delta b = -6.98$ ,  $\Delta C =$   
 $-8.46$ ,  $\Delta H = -3.46$  y  $\Delta E = 9.56$ .



# Conclusions

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This study compares the kinetics of solar drying of epazote in a direct solar dryer made of commonly used materials with another that uses a novel material for these applications: titanium oxide.

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In addition to matching the times obtained with the polycarbonate dryer, the final moisture content is lower.

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The final water activity was the same in both cases.

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Additionally, the highest drying rates are observed in the SIT.

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However, the most important advantage found in SIT is that it provides a much higher quality product, according to the colorimetry results that are observed.

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The smallest difference in color compared to fresh epazote is correlated with the preservation of the desired medicinal compounds in the epazote.

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Finally, it was determined that the model that best represents both kinetics is the Weibull.

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